



AFRL-RX-WP-JA-2015-0050

**SYNTHESIS OF FEW-LAYER, LARGE AREA
HEXAGONAL-BORON NITRIDE BY PULSED LASER
DEPOSITION (POSTPRINT)**

**Nicholas R. Glavin, Michael H. Check, and Andrey A. Voevodin
AFRL/RXAN**

**Timothy S. Fisher
School of Mechanical Engineering and Birck Nanotechnology Center**

**Michael L. Jespersen and Jianjun Hu
University of Dayton Research Institute**

**Al M. Hilton
Wyle Laboratories**

**SEPTEMBER 2014
Interim Report**

Distribution A. Approved for public release; distribution unlimited.

See additional restrictions described on inside pages

STINFO COPY

**AIR FORCE RESEARCH LABORATORY
MATERIALS AND MANUFACTURING DIRECTORATE
WRIGHT-PATTERSON AIR FORCE BASE, OH 45433-7750
AIR FORCE MATERIEL COMMAND
UNITED STATES AIR FORCE**

NOTICE AND SIGNATURE PAGE

Using Government drawings, specifications, or other data included in this document for any purpose other than Government procurement does not in any way obligate the U.S. Government. The fact that the Government formulated or supplied the drawings, specifications, or other data does not license the holder or any other person or corporation; or convey any rights or permission to manufacture, use, or sell any patented invention that may relate to them.

This report was cleared for public release by the USAF 88th Air Base Wing (88 ABW) Public Affairs Office (PAO) and is available to the general public, including foreign nationals.

Copies may be obtained from the Defense Technical Information Center (DTIC)
(<http://www.dtic.mil>).

AFRL-RX-WP-JA-2015-0050 HAS BEEN REVIEWED AND IS APPROVED FOR
PUBLICATION IN ACCORDANCE WITH ASSIGNED DISTRIBUTION STATEMENT.

//Signature//

NICHOLAS R. GLAVIN
Nanoelectronic Materials Branch
Functional Materials Division

//Signature//

DIANA M. CARLIN, Chief
Nanoelectronic Materials Branch
Functional Materials Division

//Signature//

TIMOTHY J. BUNNING, Chief
Functional Materials Division
Materials and Manufacturing Directorate

This report is published in the interest of scientific and technical information exchange, and its publication does not constitute the Government's approval or disapproval of its ideas or findings.

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 074-0188	
Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing this collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Defense, Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number. PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.				
1. REPORT DATE (DD-MM-YYYY) September 2014		2. REPORT TYPE Interim		3. DATES COVERED (From – To) 29 January 2013 – 27 August 2014
4. TITLE AND SUBTITLE SYNTHESIS OF FEW-LAYER, LARGE AREA HEXAGONAL-BORON NITRIDE BY PULSED LASER DEPOSITION (POSTPRINT)			5a. CONTRACT NUMBER FA8650-11-D-5401-0011	
			5b. GRANT NUMBER	
			5c. PROGRAM ELEMENT NUMBER 62102F	
6. AUTHOR(S) (see back)			5d. PROJECT NUMBER 4348	
			5e. TASK NUMBER	
			5f. WORK UNIT NUMBER X0K9	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) (see back)			8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES) Air Force Research Laboratory Materials and Manufacturing Directorate Wright Patterson Air Force Base, OH 45433-7750 Air Force Materiel Command United States Air Force			10. SPONSOR/MONITOR'S ACRONYM(S) AFRL/RXAN	
			11. SPONSOR/MONITOR'S REPORT NUMBER(S) AFRL-RX-WP-JA-2015-0050	
12. DISTRIBUTION / AVAILABILITY STATEMENT Distribution A. Approved for public release; distribution unlimited. This report contains color.				
13. SUPPLEMENTARY NOTES PA Case Number: 88ABW-2014-2367, Clearance Date: 16 May 2014. Journal article published in Thin Solid Films 572 (2014) 245-250. The U.S. Government is joint author of the work and has the right to use, modify, reproduce, release, perform, display or disclose the work. The final publication is available at http://dx.doi.org/10.1016/j.tsf.2014.07.059 .				
14. ABSTRACT Pulsed laser deposition (PLD) has been investigated as a technique for synthesis of ultra-thin, few-layer hexagonal boron nitride (h-BN) thin films on crystalline highly ordered pyrolytic graphite (HOPG) and sapphire (0001) substrates. The plasma-based processing technique allows for increased excitations of deposited atoms due to background nitrogen gas collisional ionizations and extended resonance time of the energetic species presence at the condensation surface. These processes permit growth of thin, polycrystalline h-BN at 700 °C, a much lower temperature than that required by traditional growth methods. Analysis of the as-deposited films reveals epitaxial-like growth on the nearly lattice matched HOPG substrate, resulting in a polycrystalline h-BN film, and amorphous BN (α -BN) on the sapphire substrates, both with thicknesses of 1.5–2 nm. Stoichiometric films with boron-to-nitrogen ratios of unity were achieved by adjusting the background pressure within the deposition chamber and distance between the target and substrate. The reduction in deposition temperature and formation of stoichiometric, large-area h-BN films by PLD provide a process that is easily scaled-up for twodimensional dielectric material synthesis and also present a possibility to produce very thin and uniform α -BN.				
15. SUBJECT TERMS hexagonal-boron nitride, nanoelectronics, two dimensional materials				
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT SAR	18. NUMBER OF PAGES 10
a. REPORT Unclassified	b. ABSTRACT Unclassified	c. THIS PAGE Unclassified		
			19a. NAME OF RESPONSIBLE PERSON (Monitor) Nicholas R. Glavin	
			19b. TELEPHONE NUBER (include area code) (937) 255-6977	

REPORT DOCUMENTATION PAGE Cont'd

6. AUTHOR(S)

Nicholas R. Glavin, Michael H. Check, and Andrey A. Voevodin - Materials and Manufacturing Directorate, Air Force Research Laboratory, Functional Materials Division
Timothy S. Fisher - School of Mechanical Engineering and Birck Nanotechnology Center
Michael L. Jespersen and Jianjun Hu - University of Dayton Research Institute
Al M. Hilton - Wyle Laboratories

7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)

AFRL/RXAN
Air Force Research Laboratory
Materials and Manufacturing Directorate
Wright-Patterson Air Force Base, OH 45433-7750

School of Mechanical Engineering and Birck Nanotechnology Center
Purdue University
West Lafayette, IN 47907

University of Dayton Research Institute
300 College Park
Dayton, OH 45469

Wyle Laboratories
Dayton, OH 45433



Synthesis of few-layer, large area hexagonal-boron nitride by pulsed laser deposition



Nicholas R. Glavin^{a,b,*}, Michael L. Jespersen^{a,c}, Michael H. Check^a, Jianjun Hu^{a,c}, Al M. Hilton^{a,d}, Timothy S. Fisher^b, Andrey A. Voevodin^a

^a Nanoelectronic Materials Branch, Air Force Research Laboratory, Wright-Patterson AFB, Dayton, OH 45433, USA

^b School of Mechanical Engineering and Birk Nanotechnology Center, Purdue University, West Lafayette, IN 47907, USA

^c University of Dayton Research Institute, 300 College Park, Dayton, OH 45469, USA

^d Wyle Laboratories, Dayton, OH 45433, USA

ARTICLE INFO

Available online 27 August 2014

Keywords:

Hexagonal-boron nitride

Nanoelectronics

Two dimensional materials

ABSTRACT

Pulsed laser deposition (PLD) has been investigated as a technique for synthesis of ultra thin, few layer hexagonal boron nitride (*h* BN) thin films on crystalline highly ordered pyrolytic graphite (HOPG) and sapphire (0001) substrates. The plasma based processing technique allows for increased excitations of deposited atoms due to background nitrogen gas collisional ionizations and extended resonance time of the energetic species presence at the condensation surface. These processes permit growth of thin, polycrystalline *h* BN at 700 °C, a much lower temperature than that required by traditional growth methods. Analysis of the as deposited films reveals epitaxial like growth on the nearly lattice matched HOPG substrate, resulting in a polycrystalline *h* BN film, and amorphous BN (*a* BN) on the sapphire substrates, both with thicknesses of 1.5–2 nm. Stoichiometric films with boron to nitrogen ratios of unity were achieved by adjusting the background pressure within the deposition chamber and distance between the target and substrate. The reduction in deposition temperature and formation of stoichiometric, large area *h* BN films by PLD provide a process that is easily scaled up for two dimensional dielectric material synthesis and also present a possibility to produce very thin and uniform *a* BN.

Published by Elsevier B.V.

1. Introduction

Two dimensional (2D) hexagonal boron nitride (*h* BN) has been investigated for use in nanotechnology applications including high speed transistor devices [1,2], flexible electronics [3,4], and high temperature coatings [5,6]. Structurally similar to graphene, *h* BN nanosheets consist of alternating B and N atoms arranged in a sp^2 bonded hexagonal lattice structure in plane, with van der Waals interactions between the layered sheets. The *h* BN sheets exhibit desirable properties including a wide band gap, high temperature stability, and inertness to many chemical environments. Boron nitride is a complementary dielectric material to graphene in 2D nanoelectronic devices as it provides an atomically smooth surface free of dangling bonds, high surface optical phonon modes, a good tunneling barrier for electrons, and minimal sites for adsorbed surface impurities [7]. The use of exfoliated *h* BN as a substrate dielectric material in graphene devices has been shown to markedly increase electron mobility in graphene by several times when compared to traditional dielectric materials such as thermally grown silicon dioxide [8].

Current methods for preparation of *h* BN nanosheets include mechanical and chemical exfoliation [9–12], high temperature chemical vapor deposition (CVD) [13–16], and PVD techniques including reactive sputtering followed by high temperature annealing [17] and pulsed laser deposition (PLD) of boron nitride nanosheets [18]. Mechanically exfoliated *h* BN flakes have been used for device level testing of numerous configurations incorporating graphene and other two dimensional materials, but the device size is limited to the size of the flake and the exfoliation thickness is not well predictable, which hinders scale up for device fabrication. Most of the CVD routes for *h* BN growth include the use of metal substrates or catalysts. Only recently, a direct growth of polycrystalline few layer *h* BN on sapphire and silicon substrates using CVD techniques was successfully demonstrated with a deposition of polyborazylene film at moderate temperatures and subsequent conversion to *h* BN by annealing at 1000 °C [16]. This also includes recent reports on the direct CVD growth of *h* BN on the surface of epitaxial graphene with an ammonia borane precursor in a vertical tube furnace at 1075 °C [19].

Clearly, one major challenge for large area, 2D *h* BN material synthesis is the reduction of the growth or annealing temperatures. Alternatively, growth of 2D materials by means of physical vapor deposition (PVD) may offer such a possibility, as these processes involve energetic plasmas which can add to the surface adatom mobilities and create

* Corresponding author at: 3005 Hobson Way, WPAFB, OH 45433-7707.

conditions for *h* BN growth at lower synthesis temperatures. The attempts with reactive sputtering [17] are promising but yielded amorphous materials which were converted to crystalline BN with the 850 °C anneal, similar to the CVD routes described above.

Pulsed laser deposition (PLD) is an alternative PVD technique to reactive sputtering, where a high energy, focused laser beam pulse causes rapid heating on the face of the target, causing both ionized and neutral components to accelerate from the target in the form of a plasma plume normal to the target surface. The plasma plume propagates in the direction of the substrate and can be thermalized by collisions with the background gas, depending on the pressure and target to substrate distance. The resulting film quality and composition on the substrate can be affected by parameters including laser wavelength, power, pulse rate, pulse duration, target to substrate working distance and relative orientation, background gas type and pressure, and substrate temperature. Whereas alternative PVD techniques present difficulties in preparing complex stoichiometric films due to differing partial pressures of the individual species in the vapor phase, PLD has proven useful in preparing complex thin films due to the rapid, localized heating, and relatively stoichiometric plasmas expelled from the target. This technique has also proven viable for growth of thicker boron nitride films of different crystalline orientations including hexagonal and cubic phases [20], at relatively low temperatures of 650 °C with B:N ratios of approximately one. Nanoscale thicknesses of *h* BN were synthesized using a high powered CO₂ pulsed laser and showed a possibility of a reduction of substrate temperature for deposition to around 400 °C, and formation of overlapping boron nitride nanosheets [18,21]. The nanosheets, however, were non uniform in thickness and morphology over large areas, limiting their ability to be used for electronic applications. Here, growth of ultra thin (less than 2 nm) stoichiometric, crystalline hexagonal phase boron nitride by means of pulsed laser deposition is discussed.

2. Experimental

Few layer boron nitride films were deposited in an ultra high vacuum chamber with residual base pressures of 2×10^{-9} Torr. Ablation from an amorphous boron nitride target (99.99% pure) was carried out using a Lambda Physik LPX300 248 nm KrF excimer laser at pulse energies of 900 mJ/pulse, 30 ns duration, and 1 Hz repetition rate. A focusing lens reduced the laser spot size to an area of $2.66 \text{ mm} \times 1.50 \text{ mm}$ with a flux of 22.5 J/cm^2 upon impact with the target. A 1000 W Astex ECR power supply supplied heat to the substrate by means of joule heating a tungsten filament to a sample temperature of 700 °C, calibrated using a thermocouple spot welded on the block just beneath the substrate. After heating each sample to the appropriate temperature, the target was cleaned by laser ablation to remove any adsorbed impurities on the target surface, while the deposition sample stage was covered with a shutter. Depositions were performed in nitrogen background gas, where pressure was controlled by a butterfly valve to preset values within 10^{-4} to 50 mTorr range and keeping the flow of ultra high purity nitrogen gas (99.9999%) at 30 sccm. After deposition, the substrate was allowed to cool to room temperature before removal from the vacuum chamber. A schematic of the deposition chamber and process is shown in Fig. 1. Two different target to substrate working distances of 1 cm and 3 cm were used in the study, with the plume at 45° to the substrate surface. Films were grown for 1 s at the 1 cm working distance, and 5 s at 3 cm working distance, in order to achieve the same thickness of BN films.

To guide the PLD process toward the formation of *h* BN films at reduced substrate temperatures, the ablated plasma dynamic was captured by collecting images of spatial, broadband plasma emission intensities timed from the laser impact on the target. A high speed Princeton Instruments 576 S/RBE ICCD camera with a 2 ns resolution in gate control was used to collect distributions of emitting species timed to the laser pulse with 100 ns steps and 100 ns gate widths. Plasma images were recorded for all visible emission within 300–900 nm

(limited by window and camera bandwidths) with no image averaging, and each image corresponding to a single laser shot.

Highly ordered pyrolytic graphite (HOPG) and sapphire samples with (0001) surface termination and dimensions $5 \text{ mm} \times 5 \text{ mm}$ provided atomically smooth surfaces with different lattice constants and electrical properties for deposition substrates. Structurally, HOPG (*a* axis lattice constant of 0.246 nm) is very similar to *h* BN, with a lattice mismatch of less than 2% and similar interlayer spacing. In order to create a clean sp^2 bonded carbon face on the substrate, exfoliation of several layers of material occurred immediately prior to placing in the deposition chamber. Polished sapphire (0001) substrates provide an ultra smooth, highly crystalline, insulating interface and are used as a substrate material in many transistor fabrications. While *c* axis oriented sapphire also provides a hexagonal planar template, the sapphire in plane *a* axis lattice constant is 0.476 nm, which is almost twice that of *h* BN (0.250 nm). Heating the substrate to 700 °C during deposition eliminated all but a minimal amount of adsorbed oxygen contamination on both substrates.

In order to assess film composition, X ray photoelectron spectroscopic (XPS) analysis was carried out using a Kratos AXIS Ultra spectrometer with a monochromatic Al K α x ray source (1486.7 eV) operated at 120 W (12 kV, 10 mA) and at approximately 4×10^{-9} Torr. In this system, the analyzer is oriented at 90° relative to the sample surface. Survey scans were collected over the binding energy range of –5 to 1200 eV, in 1 eV steps, using a dwell time of 300 ms and an analyzer pass energy of 160 eV. High resolution spectra of the B 1s, N 1s, O 1s, and C 1s regions were acquired using an energy step size of 0.1 eV, a dwell time of 500 ms, and an analyzer pass energy of 20 eV. Since BN is an insulator, a low energy electron flood source was utilized for charge compensation in cases where differential surface charging was observed. XPS spectra were analyzed using the CasaXPS software. The binding energy scale was calibrated to the C 1s peak from adventitious carbon at 284.8 eV. The atomic concentration of oxygen atoms was corrected by subtracting the O 1s contributions from adventitious carbon contamination and from the underlying sapphire substrate (O:Al for clean sapphire measured on this spectrometer was approximately 1.5) [22]. Peak areas were determined using a Shirley background subtraction, and atomic concentrations were calculated by applying relative sensitivity factors specific to the XPS instrument. Curve fits for the B 1s and N 1s peaks were obtained by fitting the peaks with a Gaussian Lorentzian function with a 30% Lorentz contribution.

To determine structural and electrical properties, the films were examined with Raman spectroscopy, AFM topography, TEM, and conductive atomic force microscopy (C-AFM). Raman structural measurements

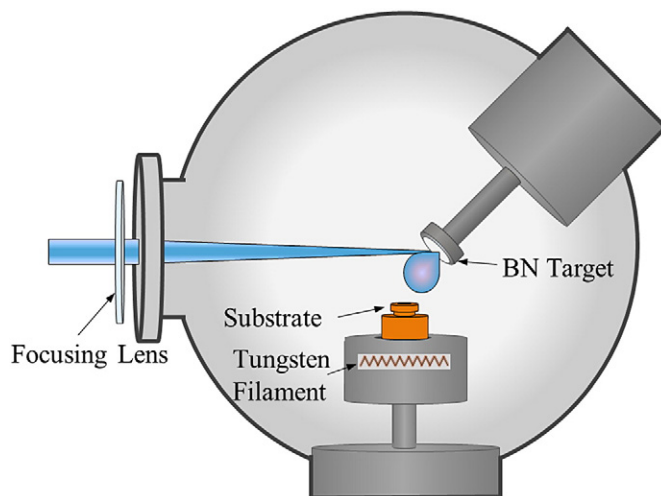


Fig. 1. Schematic of PLD setup.

were performed using a Renishaw Raman system with an Ar ion laser at 514 nm and 8 mW input power. AFM topographical images of *h* BN on HOPG and sapphire were acquired using a Bruker Dimension Icon AFM. For these measurements, the AFM was operated in tapping mode using an AFM tip with a nominal radius of curvature of ~ 2 nm (SNL 10, Bruker). Conductive atomic force microscope (CAFM) measurements were used to verify large area and pin hole free *h* BN films grown on the conductive HOPG substrate, and to determine tunneling performance at various locations on the film. Scans were taken using a doped diamond probe (DDESP 10, Bruker) to avoid tip wear, and electrical current between the biased tip and the grounded HOPG substrate was recorded using a current amplifier having a sensitivity of 1 nA/V.

3. Results and discussion

Altering the working distance as well as changing the background gas pressure within the chamber strongly impacts the incoming plasma energies and local composition in the PLD process. In particular, extending the working distance and increasing background pressure maximize the probability of collisions between ablated plasma species and background gas, reducing the incoming plasma energies by thermalization, and allowing for a longer reaction time of the energetic ablated species with the background gas species [23]. Depending on the laser energy and target material, there is an optimum pressure–distance parameter space beyond which the plumes will be fully thermalized without reaching the substrate location, and the benefits of the energetic plasma growth as compared to a CVD will be lost.

Fig. 2 shows a series of images taken at 3 cm target to substrate working distance, depicting broadband plasma emission intensities as a function of time and background pressure of ultra high purity nitrogen gas. At 600 ns after the laser impacts the target, the plume dynamics appear to be fairly similar across the range of pressures examined in this study (10^{-4} mTorr, 5 mTorr, and 50 mTorr). Significant differences are evident at 1200 ns, when the plume is spatially confined due to collisions with other gas molecules inside of the plume and at the plume

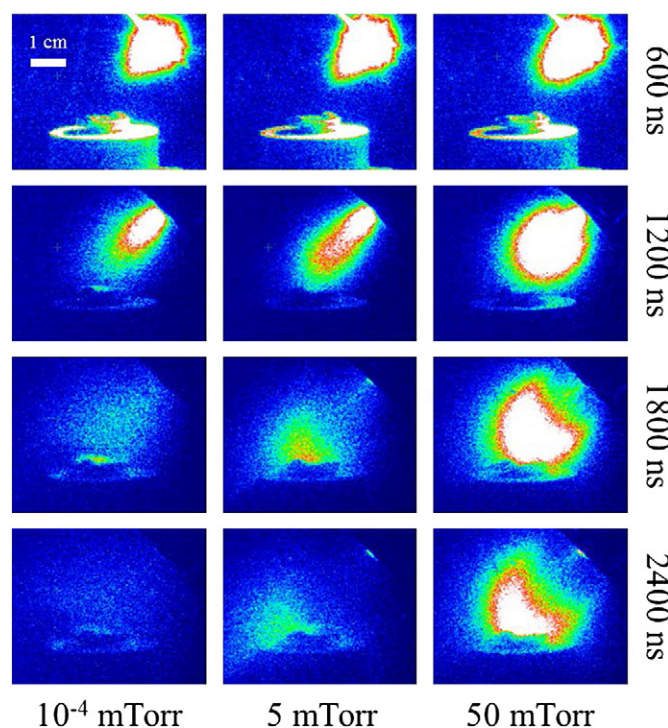


Fig. 2. CCD camera image of plasma emission intensities for 3 cm target-to-substrate working distance at various pressures and time after impact of the laser pulse on the target.

boundary. At 1800 ns and 2400 ns, plasma confinement is observed at 50 mTorr, in contrast to the diffuse, unconfined plume that has nearly dissipated entirely at the same times at 10^{-4} mTorr. Such plasma confinement is well known for plasma conditions leading to the formation of clusters and radical species from both target atoms and reactive background gas [24].

Furthermore, Fig. 2 clearly shows the additional excitations of the energetic plasma plume with the substrate. This additional plasma excitation at the substrate surface lasts for relatively long times and with a much stronger intensity when the background pressure is increased. The energetic plume lasted as long as 4 μ s after the laser pulse at a pressure of 50 mTorr. Such conditions create the extended presence of chemically reactive species (e.g. atomic boron and nitrogen), and possibly their radicals with increased vibrational temperatures at the film condensation surface. This is supported by earlier reports on PLD process optimization for a low temperature growth of ceramic and fullerene structures, where plume collisions with the substrates provided reactive species and high vibration energy radicals for increased adatom mobility, extended surface diffusion times, and reactivity needed for the formation of crystalline films at low substrate temperatures [25,26]. For the shorter working distances, similar plasma plume evolution processes were observed as in Fig. 2, but plumes had much less time for collisional reactions with the background gas before reaching the substrate surface.

Representative high resolution spectra of the B1s and N1s regions are shown in Fig. 3a and b. The B 1s peak is fit with two components at 190.6 eV and 192.1 eV, corresponding to BN and BN_xO_y , respectively. The small amount of BN_xO_y observed as a shoulder on the B 1s spectra is seen in other polycrystalline *h* BN films [16], and is most likely due to exposure to ambient conditions and oxidation at the polycrystalline

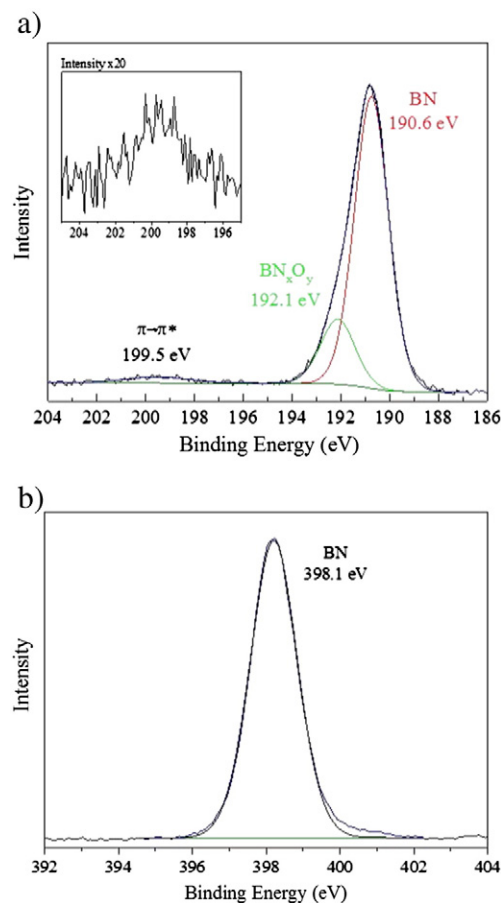


Fig. 3. a) Representative XPS scan of the B 1s peak as well as the $\pi \rightarrow \pi^*$ satellite shifted 9 eV from the B 1s peak, and b) a representative scan of the N 1s peak.

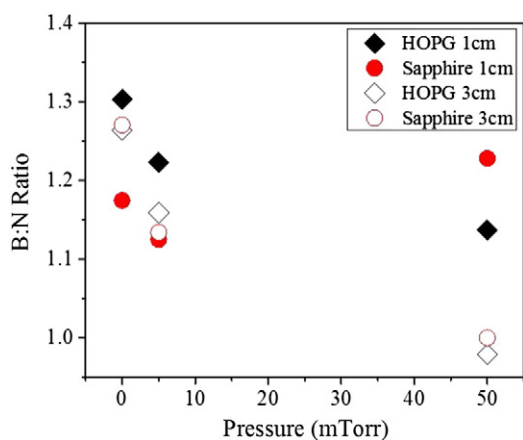


Fig. 4. Boron to nitrogen ratio as a function of nitrogen background pressure within the deposition chamber.

grain edges. The N 1s peak is fit with a single component at 398.1 eV. High resolution spectra of the B 1s region for films grown on HOPG also revealed a broad satellite peak at 9 eV higher binding energy from the main B 1s line [27]. This energy difference is characteristic of the $\pi \rightarrow \pi^*$ transition in *h* BN and was only observed for the films grown on HOPG. The crystalline formation on only the lattice matched sp^2 bonded carbon phase indicates an epitaxial like growth phenomena, where the incoming flux of deposited species has sufficient energy and surface diffusion time to restructure and match the structural template of the substrate material at the condensation surface.

Fig. 4 displays the boron to nitrogen ratios, calculated from the XPS data, as a function of both working distance and pressure for PLD grown thin films of boron nitride. The B:N ratio decreases with increasing pressure for all deposited films, with the exception of those prepared on sapphire at 1 cm working distance. The decrease in the ratio indicates a more stoichiometric construction of the films as the number of collisions between the plasma and the background gas increases. Independent of the substrate material, a 1:1 B:N stoichiometry is observed in films grown at greater working distance and at 50 mTorr background pressure. The interaction with the background gas at this pressure appears to optimize generation of BN at the interface between the plasma and substrate. In addition to the proper target to substrate working

distance, the increased number of collisions with background gas molecules greatly helped in the formation of stoichiometric BN thin films.

In addition to the lattice parameter difference, a sapphire (0001) surface energy of 1.2 J/m^2 is an order of magnitude larger than that of the HOPG (0001) surface, where $0.1\text{--}0.2 \text{ J/m}^2$ values are typical. From this observation, the HOPG surface with a low surface energy will allow for a longer dwelling of mobile species deposited on the surface from the ablated plasma plumes and a better opportunity for crystalline *h* BN formation. The E_{2g} shear mode, characteristic of *h* BN [17], is visible at 1362 cm^{-1} of samples grown on the HOPG substrates only, as seen in Fig. 5a. Peaks in the Raman spectra of *h* BN on HOPG include the highly ordered G, G*, and G' bands of HOPG at 1581 cm^{-1} , 2440 cm^{-1} , and 2725 cm^{-1} , respectively. The D band of highly ordered graphite (1352 cm^{-1}) is not evident in scans prior to deposition, as the substrates are very highly ordered, and does not show up post deposition. The latter is an important observation, as one of the concerns for thin film growth with PLD is a potential substrate damage and defect generation due to the energetic particle bombardment. The background gas collisional deceleration of the ablated plasma in this study had helped to reduce the kinetic energy of the arrival species to below $7\text{--}8 \text{ eV}$ de cohesion energy for carbon atoms from the HOPG surface [28]. The observed crystalline *h* BN Raman signature for the growth on the HOPG substrate further corroborates the XPS $\pi \rightarrow \pi^*$ satellite, suggesting epitaxial like growth on the lattice matched interface, which was not evident on the sapphire. Similar epitaxial like growth has been exhibited by cubic boron nitride films on the nearly latticed matched diamond (0001) surface [29].

The different working distances appear to influence the Raman response distribution of grain boundaries, as seen in Fig. 5a. The longer working distance results in a smaller FWHM for the E_{2g} Raman peak, indicating a larger average grain size within the polycrystalline material. The extremely high intensity Raman shift of the G peak results in an asymmetric *h* BN Raman response, making estimation of the FWHM difficult to determine. To correct for this, a background subtraction of the pristine HOPG wafer from the BN Raman signal was performed. The FWHM for the most crystalline sample at 3 cm WD and 50 mTorr was 58 cm^{-1} , indicating an average in plane domain size to be 2.9 nm for the most crystalline BN film grown at 700°C on an HOPG substrate, as compared to other polycrystalline BN films [30].

AFM topographical scans indicated large area film coverage and do not exhibit formation of micron sized domains of BN, as is the case in many CVD grown BN films [31,32]. The RMS roughness of the BN films

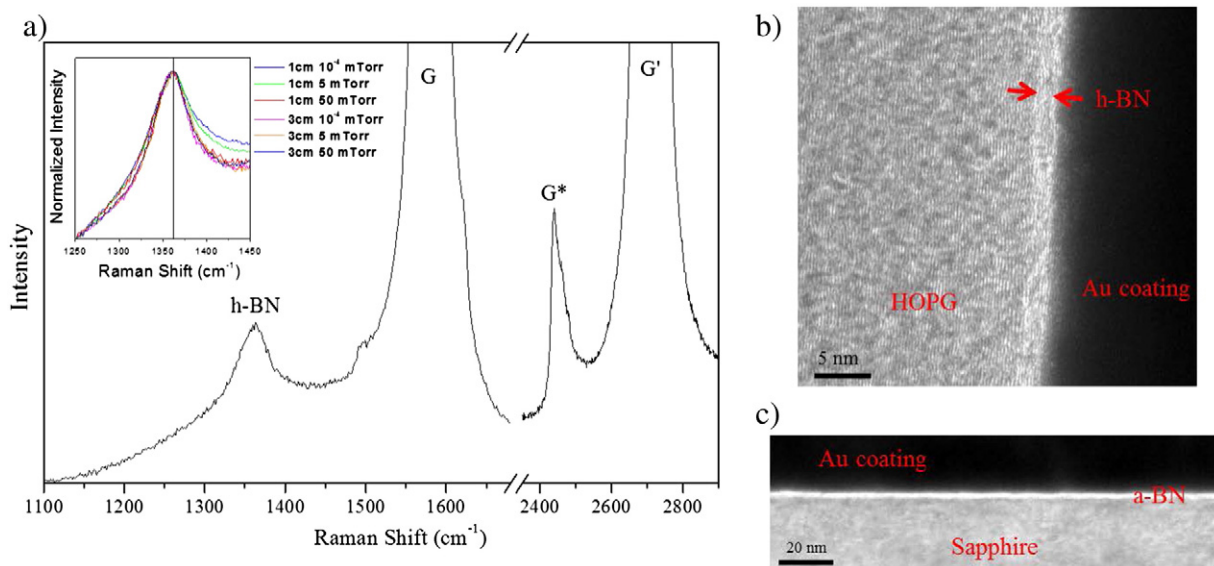


Fig. 5. a) Raman spectra from polycrystalline *h*-BN on HOPG, and the inset shows the normalized intensity of the E_{2g} shear mode as a function of nitrogen background pressure; b) cross-sectional TEM image of approximately 2 nm, polycrystalline *h*-BN on HOPG; and c) ultra-smooth amorphous boron nitride on crystalline sapphire (0001).

was independent of deposition pressure conditions, and yielded an average of 0.493 nm on HOPG and 0.133 nm on sapphire for a given $4 \mu\text{m}^2$ area.

To confirm the film thickness and crystallinity, the samples were cross sectioned and studied with transmission electron microscopy (TEM). Determination of thickness of the crystalline BN films on HOPG via TEM proved difficult due to similar e beam absorption coefficients and very close inter plane lattice spacing between of these materials. Nevertheless, the TEM image in Fig. 5b clearly indicates a surface layer, where due to the atomic mass contrast an ordered surface layer was determined as that of *h* BN, and the thickness was estimated to be 2 nm, or about 6 atomic layers. Similarly to XPS and Raman spectroscopy results, TEM studies also confirmed that amorphous BN films were grown on sapphire. The film thickness was approximately 2 nm when grown at 50 mTorr, with large area coverage and no evidence of pin holes within the amorphous film (Fig. 5c).

Complementary to TEM, *h* BN thickness estimations were performed utilizing XPS data based upon the exponential decay of substrate photoelectrons with increasing overlayer thickness due to inelastic scattering effects. Atomic concentration ratios between unique, non interfering photoelectron peaks from the overlayer (B 1s, N 1s) and the substrate (C 1s, Al 2p) were fit to a homogeneous, flat overlayer model incorporating the inelastic mean free paths (IMFPs) of substrate photoelectrons within the substrate, substrate photoelectrons within the overlayer, and overlayer photoelectrons within the overlayer in order to determine an effective overlayer thickness. This method has been detailed elsewhere [33,34]. The data confirmed film thicknesses of 1.6 nm on HOPG and 2 nm on sapphire and show that PLD offers good control over film thickness for macroscopic length scales, as the XPS analysis spot size was at $800 \mu\text{m}$ in diameter. TEM and XPS thickness measurements were repeated in multiple locations over the $5 \text{ mm} \times 5 \text{ mm}$ sample area, yielding the same results at each spot. It is known from PLD literature that the process allows for good film uniformity, and with the addition of beam rastering and sample motion, larger area samples can potentially be accommodated with similar large area uniformity as those seen in this study [35].

Fig. 6 shows eight C AFM scans on the PLD grown *h* BN films performed over the $5 \text{ mm} \times 5 \text{ mm}$ sample that indicate large area coverage free of pin holes and good electrical performance uniformity. Assuming a 1.8 nm BN film, the average low biased tunneling resistance (R_{LB}) for the polycrystalline *h* BN film is $1.811 \Omega/\text{m}$, calculated by the slope of the low bias regime, as seen in the inset of Fig. 6. The polycrystalline film will most likely exhibit a different electrical response from a pristine flake due to inherent grain boundaries within the film, resulting in slightly differing low bias I–V characteristics. Pristine *h* BN flakes have shown to have a high dependence of low biased tunneling

resistance on thickness, in the range of $0.5 \Omega/\text{m}$ for 1 nm thick flakes to $100 \Omega/\text{m}$ for 2 nm thick flakes [36].

4. Conclusions

PLD has been investigated as a new approach for the synthesis of ultra thin boron nitride films in contrast to currently used CVD processes. This technique has been shown to produce stoichiometric, polycrystalline four to six layer BN films within 5 s of deposition on 25 mm^2 areas at 700°C temperatures. This was achieved by in situ recording and adjusting pulsed plasma plume evolutions to optimize growth conditions at the condensation surfaces with the presence of reactive, thermally excited, high mobility, and long diffusion time species on the condensation surface. Growth of crystalline *h* BN on both lattice matched substrates (e.g., HOPG) as well as amorphous, smooth *a* BN films on crystalline substrates (e.g., sapphire) was demonstrated with complete coverage of the large area without pinholes, or any other cross thickness defects as confirmed with a number of techniques. When compared to traditional CVD methods, the technique opens up the possibility of preparing high quality BN films on a wide range of substrates previously incompatible with traditional BN growth methods, as well as the potential to scale the film production to large areas, with the high throughput required for commercialization.

Acknowledgments

The authors would like to gratefully acknowledge support from Art Safriet and John Bultman of the University of Dayton Research Institute.

References

- [1] A. Hsu, W. Han, C.S. Yong, B. Maillly, Z. Xu, Y. Lili, S. Yumeng, H.L. Yi, M. Dubey, K.K. Ki, K. Jing, T. Palacios, Large-area 2-D electronics: materials, technology, and devices, *Proc. IEEE* 101 (7) (2013) 1638.
- [2] K.H. Lee, H.J. Shin, J. Lee, I. y Lee, G.H. Kim, J.Y. Choi, S.W. Kim, Large-scale synthesis of high-quality hexagonal boron nitride nanosheets for large-area graphene electronics, *Nano Lett.* 12 (2) (2012) 714.
- [3] N. Guo, J. Wei, Y. Jia, H. Sun, Y. Wang, K. Zhao, X. Shi, L. Zhang, X. Li, A. Cao, H. Zhu, K. Wang, D. Wu, Fabrication of large area hexagonal boron nitride thin films for bendable capacitors, *Nano Res.* 6 (8) (2013) 602.
- [4] G.H. Lee, Y.J. Yu, X. Cui, N. Petrone, C.H. Lee, M.S. Choi, D.Y. Lee, C. Lee, W.J. Yoo, K. Watanabe, T. Taniguchi, C. Nuckolls, P. Kim, J. Hone, Flexible and transparent MoS₂ field-effect transistors on hexagonal boron nitride-graphene heterostructures, *ACS Nano* 7 (9) (2013) 7931.
- [5] A.P. Barboza, H. Chacham, C.K. Oliveira, T.F.D. Fernandes, E.H.M. Ferreira, B.S. Archanjo, R.J.C. Batista, A.B. De Oliveira, B.R.A. Neves, Dynamic negative compressibility of few-layer graphene, *h*-BN, and MoS₂, *Nano Lett.* 12 (5) (2012) 2313.
- [6] L. Changgu, L. Qunyang, W. Kalb, L. Xin-Zhou, H. Berger, R.W. Carpick, J. Hone, Frictional characteristics of atomically thin sheets, *Science* 328 (5974) (2010) 76.
- [7] L. Britnell, R.V. Gorbachev, R. Jalil, B.D. Belle, F. Schedin, M.J. Katsnelson, L. Eaves, S.V. Morozov, A.S. Mayorov, N.M.R. Peres, A.H. Castro Neto, J. Leist, A.K. Geim, L.A. Ponomarenko, K.S. Novoselov, Electron tunneling through ultrathin boron nitride crystalline barriers, *Nano Lett.* 12 (3) (2012) 1707.
- [8] W. Han, T. Taychatanapat, A. Hsu, K. Watanabe, T. Taniguchi, P. Jarillo-Herrero, T. Palacios, BN/graphene/BN transistors for RF applications, *IEEE Electron Device Lett.* 32 (9) (2011) 1209.
- [9] C.R. Dean, A.F. Young, I. Meric, C. Lee, L. Wang, S. Sorgenfrei, K. Watanabe, T. Taniguchi, P. Kim, K.L. Shepard, J. Hone, Boron nitride substrates for high-quality graphene electronics, *Nat. Nanotechnol.* 5 (10) (2010) 722.
- [10] N. Alem, R. Erni, C. Kisielowski, M.D. Rossell, W. Gannett, A. Zettl, Atomically thin hexagonal boron nitride probed by ultrahigh-resolution transmission electron microscopy, *Phys. Rev. B Condens. Matter Mater. Phys.* 80 (15) (2009) 155425.
- [11] A.G.F. Garcia, M. Neumann, F. Amet, J.R. Williams, K. Watanabe, T. Taniguchi, D. Goldhaber-Gordon, Effective cleaning of hexagonal boron nitride for graphene devices, *Nano Lett.* 12 (9) (2012) 4449.
- [12] Y. Wang, Z. Shi, J. Yin, Boron nitride nanosheets: large-scale exfoliation in methanesulfonic acid and their composites with polybenzimidazole, *J. Mater. Chem.* 21 (30) (2011) 11371.
- [13] L. Song, L. Ci, H. Lu, P.B. Sorokin, C. Jin, J. Ni, A.G. Kvashnin, D.G. Kvashnin, J. Lou, B.J. Yakobson, P.M. Ajayan, Large scale growth and characterization of atomic hexagonal boron nitride layers, *Nano Lett.* 10 (8) (2010) 3209.
- [14] A. Ismach, H. Chou, D.A. Ferrer, Y. Wu, S. McDonnell, H.C. Floresca, A. Covacevich, C. Pope, R. Piner, M.J. Kim, R.M. Wallace, L. Colombo, R.S. Ruoff, Toward the controlled synthesis of hexagonal boron nitride films, *ACS Nano* 6 (7) (2012) 6378.

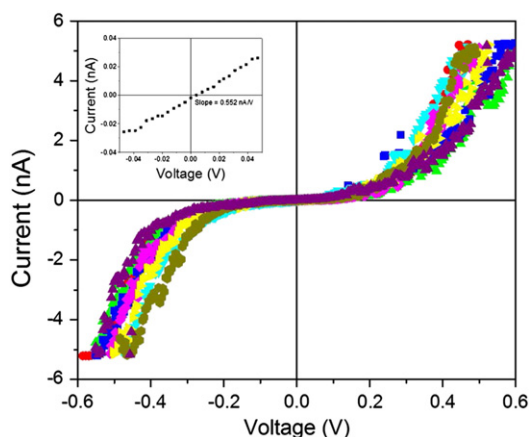


Fig. 6. Conducting AFM I–V curves of eight spots on a $5 \text{ mm} \times 5 \text{ mm}$ BN-coated HOPG sample, and the inset shows the average value of the low-biased tunneling curve.

- [15] Y. Shi, C. Hamsen, X. Jia, K.K. Kim, A. Reina, M. Hofmann, A.L. Hsu, K. Zhang, H. Li, Z.Y. Juang, M. Dresselhaus, L.J. Li, J. Kong, Synthesis of few-layer hexagonal boron nitride thin film by chemical vapor deposition, *Nano Lett.* 10 (10) (2010) 4134.
- [16] M.S. Bresnehan, M.J. Hollander, M. Wetherington, K. Wang, T. Miyagi, G. Pastir, D.W. Snyder, J.J. Gengler, A.A. Voevodin, W.C. Mitchel, J.A. Robison, Prospects of direct growth boron nitride films as substrates for graphene electronics, *J. Mater. Res.* 29 (3) (2014) 459–471.
- [17] P. Sutter, J. Lahiri, P. Zahl, B. Wang, E. Sutter, Scalable synthesis of uniform few-layer hexagonal boron nitride dielectric films, *Nano Lett.* 13 (1) (2013) 276.
- [18] M. Sajjad, G. Morell, P. Feng, Advance in novel boron nitride nanosheets to nanoelectronic device applications, *ACS Appl. Mater. Interfaces* 5 (11) (2013) 5051.
- [19] Y.C. Lin, N. Lu, N. Perea-Lopez, J. Li, Z. Lin, X. Peng, C.H. Lee, C. Sun, L. Calderin, P.N. Browning, M.S. Bresnehan, M.J. Kim, T.S. Mayer, M. Terrones, J.A. Robinson, Direct synthesis of van der Waals solids, *ACS Nano* 8 (4) (2014) 3715–3723.
- [20] N. Acacia, E. Fazio, F. Neri, P.M. Ossi, S. Trusso, N. Santo, Pulsed laser deposition of boron nitride thin films, *Radiat. Eff. Defects Solids* 163 (4–6) (2008) 293.
- [21] M. Sajjad, H.X. Zhang, X.Y. Peng, P.X. Feng, Effect of substrate temperature in the synthesis of BN nanostructures, *Phys. Scr.* 83 (6) (2011) 065601.
- [22] B.P. Payne, M.C. Beisinger, N.S. McIntyre, X-ray photoelectron spectroscopy studies of reactions on chromium metal and chromium oxide surfaces, *J. Electron Spectrosc. Relat. Phenom.* 184 (1–2) (2011) 29.
- [23] Pulsed Laser Deposition of Thin Films: Applications-Led Growth of Functional Materials, Wiley-Interscience, 2007.
- [24] D.H. Lowndes, D.B. Geohegan, A.A. Puretzky, D.P. Norton, C.M. Rouleau, Synthesis of novel thin-film materials by pulsed laser deposition, *Science* 273 (5277) (1996) 898.
- [25] A.A. Voevodin, J.G. Jones, J.S. Zabinski, Characterization of ZrO₂/Y₂O₃ laser ablation plasma in vacuum, oxygen, and argon environments, *Journal of Applied Physics* 88 (2000) 1088.
- [26] A.A. Voevodin, J.G. Jones, J.S. Zabinski, L. Hultman, Plasma characterization during laser ablation of graphite in nitrogen for the growth of fullerene-like CN_x films, *J. Appl. Phys.* 92 (2002) 724.
- [27] R. Trehan, Y. Lifshitz, J.W. Rabalais, Auger and X-ray electron spectroscopy studies of hBN, cBN, and N₂⁺ ion irradiation of boron and boron nitride, *J. Vac. Sci. Technol.* 8 (6) (1990) 4026.
- [28] N. Ooi, A. Rairkar, J.B. Adams, Density functional study of graphite bulk and surface properties, *Carbon* 44 (2) (2006) 231.
- [29] X.W. Zhang, H.G. Boyen, N. Deyneka, P. Ziemann, F. Banhart, M. Schreck, Epitaxy of cubic boron nitride on (001) oriented diamond, *Nat. Mater.* 2 (5) (2003) 312.
- [30] R.J. Nemanich, S.A. Solin, R.M. Martin, Light scattering study of boron nitride microcrystals, *Phys. Rev. B Condens. Matter* 23 (12) (1981) 6348.
- [31] R.Y. Tay, X. Wang, S.H. Tsang, G.C. Loh, R.S. Singh, H. Li, G. Mallick, E.H. Tong Teo, A systematic study of the atmospheric pressure growth of large-area hexagonal crystalline boron nitride film, *J. Mater. Chem. C* 2 (9) (2014) 1650.
- [32] K.K. Kim, A. Hsu, X. Jia, S.M. Kim, Y. Shi, M. Hofmann, D. Nezich, J.F. Rodriguez-Nieva, M. Dresselhaus, T. Palacios, J. Kong, Synthesis of monolayer hexagonal boron nitride on Cu foil using chemical vapor deposition, *Nano Lett.* 12 (1) (2012) 161.
- [33] XPS Thickness Solver, version, <https://nanohub.org/resources/xpsts2012>.
- [34] C.J. Powell, A. Jablonski, NIST electron inelastic-mean-free-path database. [Version 1.2], National Institute of Standards and Technology, Gaithersburg, MD, 2010.
- [35] Jim Greer, Large-area commercial pulsed laser deposition, in: Robert Eason (Ed.), Pulsed Laser Deposition of Thin Films: Applications-led Growth of Functional Materials, John Wiley & Sons, Inc, 2007, p. 191.
- [36] G.H. Lee, Y.J. Yu, C. Lee, C. Dean, K.L. Shepard, P. Kim, J. Hone, Electron tunneling through atomically flat and ultrathin hexagonal boron nitride, *Appl. Phys. Lett.* 99 (24) (2011).